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## PRELIMINARY NOTE

Fluoroalkylbenzenes: Synthesis of (S)-2,3-Dimethoxy-5-[(3-fluoro)propyl]-6-hydroxy-N-[(1-ethyl-2-pyrrolidinyl)methyl]benzamide

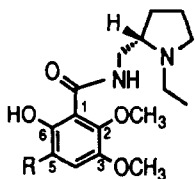
John M. GERDES\*, John E. BISHOP and Chester A. MATHIS

Research Medicine and Radiation Biophysics, Lawrence Berkeley Laboratory, University of California, Berkeley, CA 94702 (U.S.A.)

## SUMMARY

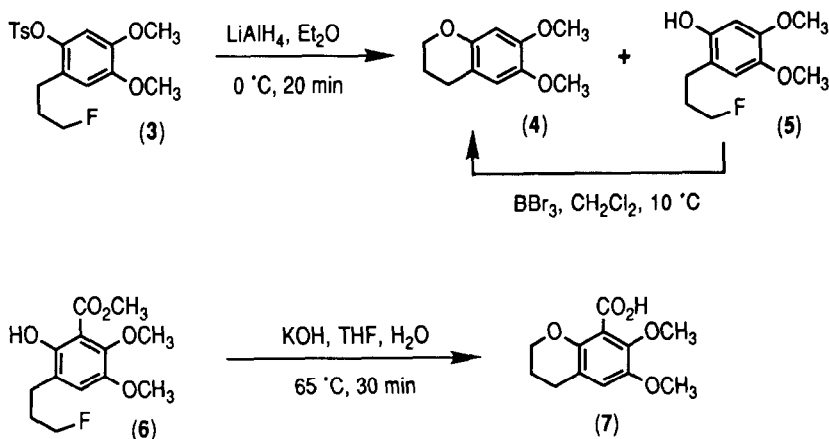
The synthesis of the titled fluoropropylsalicylamide (2) was achieved by a route which avoided the intramolecular cyclization reactions of *ortho*-[(3-fluoro)propyl]phenol substrates.

The study of fluorinated analogues of biologically important molecules [1] including derivatives in which aromatic halogen substituents have been replaced with fluoroalkyl groups [2] and the recent report that the *ortho*-halogenated bromosalicylamide (1) functions as a potent dopamine D-2 receptor antagonist [3] makes the fluoropropylsalicylamide (2) a logical and highly desirable synthetic target for *in vivo* and *in vitro* studies [2]. The novel *ortho*-[(3-fluoro)propyl]phenol structural feature of (2) is without literature precedent and is anticipated under certain reaction conditions to lead to loss of fluorine and intramolecular cyclization to provide benzopyran-type products [4]. Additionally, functional groups of penta-substituted aromatic substrates are known to possess increased reactivity profiles as a result of steric crowding [1]. Thus before we could develop an efficient synthesis of (2), a preliminary investigation of the reactivity of several *ortho*-[(3-fluoro)propyl]phenols was undertaken.



(1) R = Br

(2) R = CH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>F



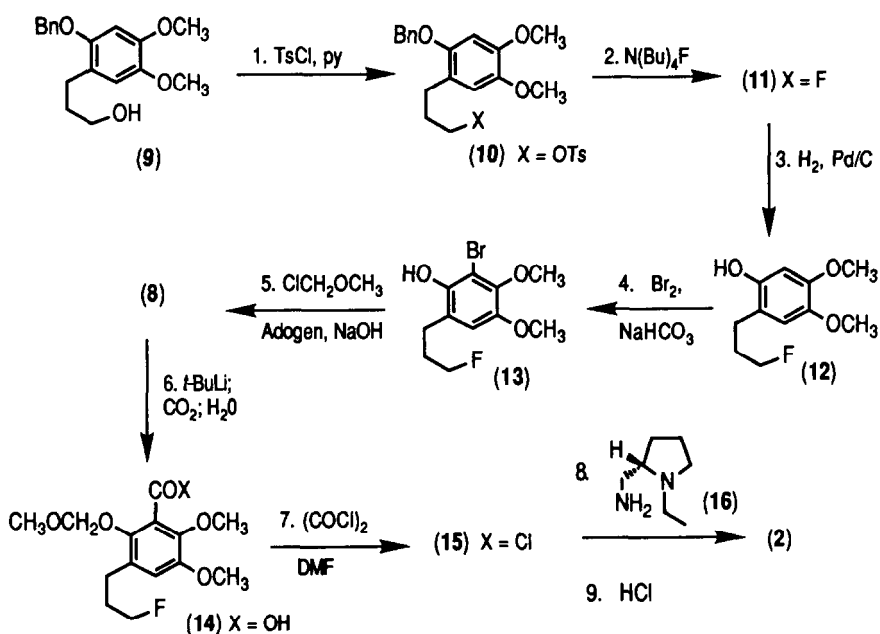
Preliminary synthetic studies revealed that *ortho*-[(3-fluoro)propyl]phenols cyclize either under Lewis acid or aqueous base reaction conditions. For example, reduction of the tosylate (3)\* with lithium aluminum hydride at 0°C afforded the cyclized benzopyran (4) (53 %) and the fluoropropylphenol (5) (34 %). The phenol (5) was converted to the pyran (4) by treatment with boron tribromide in dichloromethane at 10 °C (63 %)[ 1]. Furthermore, *ortho*-fluoropropylsalicylic ester (6) rapidly cyclized to afford the benzopyran acid (7) (76 %) when treated with potassium hydroxide in aqueous THF at 65 °C for 30 min [5]. Indeed, the labile nature of these *ortho*-[(3-fluoro)propyl]phenols to these reaction conditions required that traditional salicylamide reaction sequences [3,6] be avoided for the preparation of (2). We therefore selected intermediate (8) as the critical precursor for

the synthesis of (2). The requisite benzoic acid for amide formation could potentially be afforded from the corresponding aryl bromide by a regiospecific heteroatom-facilitated lithium-for-bromine exchange reaction followed by trapping with carbon dioxide [7], and the methoxymethyl ether of the phenol moiety could be unmasked under mild noncyclizing aqueous acid conditions.

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\* Preparation of a detailed account of the syntheses and chemistry of the new molecules reported here as well as other molecules possessing the *ortho*-[(3-fluoro)propyl]phenol structural unit is currently in progress. The spectroscopic qualities (<sup>1</sup>H and <sup>19</sup>F NMR, and IR) and elemental analyses of all new compounds are in agreement with the assigned structures.

The intermediate (**8**) was synthesized from the previously reported 3-[4,5-dimethoxy-2-(phenylmethoxy)benzene]propanol (**9**) (Scheme 1) [8]. Treatment of alcohol (**9**) with *p*-toluenesulfonyl chloride and pyridine in dichloromethane at 0 °C afforded the tosylate (**10**) (91%), which was converted to the fluoropropylbenzene (**11**) with tetrabutylammonium fluoride in refluxing tetrahydrofuran (93%). Removal of the benzyl ether protecting group by hydrogenolysis over palladium on carbon in methanol at 20 °C yielded the corresponding fluoropropylphenol (**12**) (100 %) which underwent regiospecific bromination with bromine buffered with sodium bicarbonate in carbon tetrachloride at 0 °C to afford (89 %) the penta-substituted bromophenol (**13**). Protection of the phenol group as the methoxymethyl (MOM) ether using chloromethyl methyl ether and mild phase transfer conditions [9] afforded (87%) the key intermediate (**8**). Bromine-for-lithium exchange of (**8**) with *t*-BuLi in ether at -78 °C



Scheme 1. Synthesis of Fluoroalkylsalicylamide (**2**).

and then reaction of the aryl anion with carbon dioxide in ether at -78 °C followed by water quench [7] provided (68 %) the fluoropropylbenzoic acid (**14**). Elaboration of (**14**) to the desired salicylamide (**2**) was achieved by first generating the corresponding acid chloride (**15**) with oxalyl chloride and dimethylformamide in dichloromethane at 0 °C (not isolated),

followed by treatment with (S)-N-ethyl-2-(aminomethyl)pyrrolidine (16) [7] in dichloromethane (20 °C), and then removal of the MOM ether group with aqueous methanolic HCl at 20 °C (78 %, 14→2).

The synthesis of (2) from alcohol (9) was accomplished in nine steps in an overall yield of 35 %. It is apparent that molecules with the novel *ortho*-[(3-fluoro)propyl]phenol structural unit can be obtained in synthetically useful yields by employing reaction conditions which do not favor intramolecular etherification processes. The initial screening of *in vitro* biological activity has revealed that fluoroalkylsalicylamide (2) is a potent dopamine D-2 antagonist similar to the bromosalicylamide (1). A full description biological properties of salicyclamide (2) will be published elsewhere.

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- 1 'Biomedical Aspects of Fluorine Chemistry' Ed. R. Filler and Y. Kobayashi, Elsevier Biomedical, Amsterdam, (1982).
- 2 J. M. Gerdes, C. A. Mathis and A. T. Shulgin, *Tetrahedron Lett.*, **29**, (1988) 6537.
- 3 T. Högberg, S. Bengtsson, T. de Paulis, L. Johansson, P. Strom, H. Hall and S. O. Ögren, *J. Med. Chem.*, **33**, (1990) 1155.
- 4 W. Dmowski, *Synth.*, (1983) 396.
- 5 For a discussion of fluorine as a leaving group in S<sub>N</sub>2 processes see 'Fluorine in Organic Chemistry' R.D. Chambers, Wiley, New York, (1973) Chpt. 5.
- 6 T. Högberg, S. Råmsby, S.-O. Ögren, U. Norinder, *Acta Pharm. Succ.*, **24**, (1987) 289.
- 7 H. W. Gschwend and H. R. Rodriguez, *Org. React.*, **26**, (1979) 1.
- 8 G. Büchi and P.-S. Chu, *J. Am. Chem. Soc.*, **103**, (1981) 2718.
- 9 F. R. van Heerden, J. J. van Zyl, J. H. Rall and E. V. Brandt, *Tetrahedron Lett.*, (1978) 661.